Algebraic-Matrix Calculation of Vibrational Levels of Triatomic Molecules

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We introduce an accurate and efficient algebraic technique for the computation of the vibrational spectra of triatomic molecules, of both linear and bent equilibrium geometry. The full three-dimensional potential energy surface (PES), which can be based on entirely ab initio data, is parametrized as a product Morse-cosine expansion, expressed in bond angle internal coordinates, and includes explicit interactions among the local modes. We describe the stretching degrees of freedom in the framework of a Morse-type expansion on a suitable algebraic basis, which provides exact analytical expressions for the elements of a sparse Hamiltonian matrix. Likewise, we use a cosine power expansion on a spherical harmonics basis for the bending degree of freedom. The resulting matrix representation in the product space is very sparse, and vibrational levels and eigenfunctions can be obtained by efficient diagonalization techniques. We apply this method to carbonyl sulfide, hydrogen cyanide, water, and nitrogen dioxide. When we base our calculations on high-quality PESs tuned to the experimental data, the computed spectra are in very good agreement with the observed band origins.

I. Introduction

The present-day fast development of new spectroscopic instruments and methods allows us to measure vibrational states with high accuracy even in the energy region near the molecular dissociation.^{1–3} Characterizing these new experimental data is especially important, for instance, for understanding the dynamics of chemical reactions and for analyzing spectra from remote regions (e.g., upper atmosphere and/or interstellar matter), where strong radiation along with low pressure can stabilize molecules in extraordinarily excited states, sometimes promoting unusual reactions. All energy levels depend sensitively on the detailed shape of the molecular potential energy surface (PES). Their qualitative and also quantitative description calls for the development of more and more accurate theoretical approaches.^{3–8}

The present-day status of the art program codes for the accurate calculation of vibrational spectra of small molecules has been developed in J. Tennyson's (TRIATOM,⁴ DVR3R,⁵...) and P. Jensen's (MORBID,⁶ TROVE⁷) groups. They are based on different approaches and find the best use for different kinds of small (in particular triatomic) molecules. TRIATOM focuses mainly on van der Waals complexes and employs Legendre and Laguerre polynomials as basis functions. The matrix elements are integrated numerically using a discrete variable representation (DVR) based on the Gauss-Jacobi and Gauss-Laguerre quadrature for all three internal coordinates. The resulting spectra depend on the DVR grid in a nonvariational fashion. The MORBID code is useful mainly for standard rigid molecules. It takes advantage of Morse oscillator basis functions, for which the matrix elements of pure stretching motion are known analytically. However, the bending basis functions are evaluated in the framework of a numerical integration technique, which can be a time-consuming step. Modern fully variational approaches use a finite basis representation of the vibrational space: the TROVE implementation in principle allows variational calculations to be performed for general polyatomic molecules of arbitrary structure. However, in that approach, the kinetic energy operator is represented only approximately as a power expansion in terms of internal coordinates. That approximation produces adequately accurate energy levels only rather low in energy, where the wave function has a limited spread around the molecular equilibrium position, where the power expansion is accurate.

In the present work we introduce a virtually exact variational method, which joins the main advantages of (i) simple algebraic forms of matrix elements, (ii) completeness of the basis set, and (iii) sparseness of the resulting Hamiltonian matrix. In concrete, we describe the stretching modes with a formalism based on the Morse oscillator,⁹ with some similarity to the algebraic approaches of ref 10. The main advantage of a potential expansion in terms of Morse coordinates is that one can use a quantum mechanical basis on which the matrix representation of the Hamiltonian operator is sparse and can be computed analytically using algebraic techniques based on generalized step operators. For the bending mode we choose a cosine power expansion of the potential and a quantum mechanical basis of spherical-harmonic functions. We then formulate and solve numerically the multidimensional problem of the vibrations of a polyatomic molecule in a product space of the different degrees of freedom. The sparse quality of the individual matrices representing the one-dimensional vibrations and the interaction terms carries forward to the global matrix representation of the total Hamiltonian on the product Hilbert space. The eigenvalues and eigenvectors of the resulting sparse matrix can then be obtained numerically quite efficiently by modern exact diagonalization tools. In principle, this method can be applied to molecules composed by any number of atoms, even though in practice the size of the product Hilbert space grows exponentially in the number of dimensions, which makes

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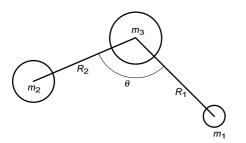


Figure 1. Scheme of internal coordinates used in the present work.

this method, like all virtually exact methods, rather unpractical for molecules with six or more atoms. In this study we focus on triatomics, but we construct the theory in terms of basic building blocks usable for successive extensions to the multidimensional potential surfaces of polyatomic molecules, which will be the subject of future investigation.

The present approach bears some resemblance to Lie algebraic methods based on a Heisenberg formulation of quantum chemistry (the second quantization of the Schrödinger equation). Several previous studies ^{10–13} use an algebraic approach where the full Hamiltonian operator is expanded in powers of \mathcal{C} , a Casimir operator of a suitable Lie algebra, for example, as follows: $\hat{H} = E_0 + \sum_{k=2}^{k} A_i [\mathcal{C}(O(2))]^i$. The trouble with that kind of expansion is that it maps to an explicit first-quantized Hamiltonian form involving intricate mixed potential and kinetic contributions, including high powers of the momentum operator and unphysical products of the momentum and position operators. To avoid such unwanted properties, we express separately the potential *V* and kinetic *T* energy operators. This can be done within a convenient scheme of ladder operators defined by means of a suitable factorization method.^{9,14}

In Section II we sketch the model used, including the Hamiltonian, basis functions for stretching and bending modes, and formulas for the matrix elements. In Section III we report and discuss the vibrational spectra of carbonyl sulfide (OCS), hydrogen cyanide (HCN), nitrogen dioxide (NO₂), and water (H₂O) as obtained with the present method. In Section IV we discuss the approximations involved in the present method and its future extensions.

II. Theory

We perform all calculations in the bond length and angle internal coordinates, as illustrated in Figure 1. This is the most suitable choice of variables for the potential expansion involving Morse functions. R_1 and R_2 represent the bond lengths between the central atom and the two end atoms, and θ is the bending angle at the central atom. Given the masses m_1 and m_2 of the end atoms and m_3 of the central atom, we define the two diatom reduced masses $\mu_1 = m_1 m_3/(m_1 + m_3)$ and $\mu_2 = m_2 m_3/(m_2 + m_3)$. The vibrational Hamiltonian, composed by a kinetic and a potential part, can be expressed in atomic units ($\hbar = m_e = q_e$ = 1) as follows:

$$H(R_1, R_2, \theta) = T(R_1, R_2, \theta) + V(R_1, R_2, \theta)$$
(1)

The standard expression for the pure vibrational kinetic energy operator in internal coordinates^{15,16} is

$$T(R_{1}, R_{2}, \theta) = -\frac{1}{2\mu_{1}} \frac{\partial^{2}}{\partial R_{1}^{2}} - \frac{1}{2\mu_{2}} \frac{\partial^{2}}{\partial R_{2}^{2}} - \frac{1}{2} \left(\frac{1}{2\mu_{1}R_{1}^{2}} + \frac{1}{\mu_{2}R_{2}^{2}} - \frac{2\cos\theta}{m_{3}R_{1}R_{2}}\right) \left(\frac{\partial^{2}}{\partial\theta^{2}} + \cot\theta\frac{\partial}{\partial\theta}\right) - \frac{\sin\theta}{m_{3}R_{1}R_{2}} \frac{\partial}{\partial\theta} + \frac{\cos\theta}{m_{3}} \left(\frac{1}{R_{1}}\frac{\partial}{\partial R_{2}} + \frac{1}{R_{2}}\frac{\partial}{\partial R_{1}}\right) - \frac{\cos\theta}{m_{3}}\frac{\partial^{2}}{\partial R_{1}\partial R_{2}} + \frac{\sin\theta}{m_{3}}\frac{\partial}{\partial\theta} \left(\frac{1}{R_{1}}\frac{\partial}{\partial R_{2}} + \frac{1}{R_{2}}\frac{\partial}{\partial R_{1}}\right) - \frac{\cos\theta}{m_{3}R_{1}R_{2}} + \frac{1}{8} \left(\frac{1}{\mu_{1}R_{1}^{2}} + \frac{1}{\mu_{2}R_{2}^{2}} + \frac{2}{m_{3}R_{1}R_{2}}\right) \frac{1}{\cos^{2}(\theta/2)}\Pi_{z}^{2} \quad (2)$$

This expression is singular at four boundary regions: at $R_i \rightarrow 0$ and at the bending extremes $\theta = 0$ and $\theta = \pi$. While the former constitute no serious practical problem, since the vibrational motion avoids these regions due to the strongly repulsive nature of the potential energy function V there, the angular singular points are easily reached by the motion: in particular $\theta = \pi$ is the angular minimum energy direction in the case of linear molecules, and this may require explicit care to obtain a numerically stable algorithm. Specifically, the Π_z operator in the final row, defined in ref 16, accounts for the rotations of the molecule around an axis attached to it and coincident with the molecular axis when the molecule reaches its linear configuration $\theta = 180^{\circ}$: for our purpose Π_z can be replaced everywhere with its eigenvalue m. The final term $\propto \Pi_z^2$ describes the kinetic energy contribution of a rotational degree of freedom of bent molecules (and in this case we omit it), but it is needed to account for the fourth vibrational degree of freedom describing the $m = \pm 1$ " Π ", $m = \pm 2$ " Δ ", etc. axially rotating vibrational excitations of those molecules such as OCS and HCN, which are linear in their equilibrium geometry. For simplicity, in the main text we will stick mostly to m = 0, while the Appendix shall deal with the general algebra describing an arbitrary integer value of *m*.

We use the following parametrization of the potential energy surface:

$$V(R_1, R_2, \theta) = \sum_{k_1, k_2, k_3 = 0}^{k_1 + k_2 + k_3 \le N_c} a_{k_1 k_2 k_3} v_1(R_1)^{k_1} v_2(R_2)^{k_2} u(\theta)^{k_3} (3)$$

This power expansion is realized in terms of Morse-related functions

$$v_i(R_i) = e^{-\alpha_i(R_i - R_{i,\min})} - 1 \tag{4}$$

for the stretching degrees of freedom i = 1 or 2 and in terms of trigonometric expressions $u(\theta) = [\cos \theta - \cos \theta_{\min}]$ for the bending degree of freedom. Similar parametrizations are employed, e.g., by P. Jensen in the MORBID code;¹⁷ also D. Xie and co-workers use this Morse-cosine expansion for several molecules, with the parameters first fixed to fit an ab initio PES and later adjusted to reproduce spectroscopic experimental data.^{18,19}

As customary, to compute a variational solution of the eigenproblem corresponding to H, we resort to the expansion of the eigenfunctions of H on a product basis:

$$\Psi_{\nu}(R_1, R_2, \theta) = \sum_{j_1, j_2, j_3} c_{\nu}^{j_1 j_2 j_3} \Phi_{j_1 j_2 j_3}(R_1, R_2, \theta)$$
(5)

where v is a complete set of vibrational quantum numbers

characterizing an eigenstate of H, $c_v^{j,j_{2j_3}}$ are yet-to-be-determined expansion coefficients, and

$$\Phi_{j_1 j_2 j_3}(R_1, R_2, \theta) = \phi_{1 j_1}(R_1) \phi_{2 j_2}(R_2) Y_{j_3}(\theta)$$
(6)

The 1-dimensional basis functions $\phi_{ij_i}(R_i)$ for the stretching modes were discussed in detail in refs 9 and 14. In particular, we use here the generalized quasi-number-state basis,^{14,20} defined as follows:

$$\phi_{ij_i}(R_i) = \sqrt{\frac{\alpha_i j_i!}{\Gamma(2\sigma_i + j_i)}} y_i^{\sigma_i} e^{-y_i/2} L_{j_i}^{2\sigma_i - 1}(y_i),$$

$$j_i = 0, 1, 2, ..., \quad i = 1 \text{ or } 2 \quad (7)$$

with

$$y_i = y_i(R_i) = (2s_i + 1)e^{-\alpha_i(R_i - R_{i,\min})}$$
 (8)

Here L_n^{ρ} are generalized Laguerre polynomials, Γ is the standard Γ function generalization of the factorial,²¹ and s_i and σ_i are suitable positive parameters, whose values we discuss below.

For the angular variable we use the following basis derived from the spherical harmonics:

$$Y_{l}(\theta) = \sqrt{2\pi} Y_{l0}(\theta, \varphi) = N_{l} P_{l}(\cos \theta) \quad \text{with} \quad N_{l} = \left(l + \frac{1}{2}\right)^{1/2} (9)$$

 $[P_l(\cos \theta)$ are standard Legendre polynomials $P_l(z) \equiv (d/dz^l)(z^2 - 1)^l/(2^l l!)]$ which form a convenient orthonormal basis set over the range $0 \le \theta \le \pi$ in the measure d cos θ . This basis was used successfully by Carter and Handy.¹⁵ In that work the authors evaluated the matrix-element integrals numerically by the Gauss–Legendre quadrature, while here we use analytical expressions detailed below.

As the Hamiltonian is a sum of terms, its matrix elements are also expressed as a sum of individual terms, and each one of them is expressed as a product of operators acting on the R_1 , R_2 , and θ variables. Accordingly, the matrix elements of H are computed as sums of products of terms, each of which refers to one oscillator individually. For example, the matrix elements of one of the kinetic contributions in eq 2 are evaluated as

$$\langle \Phi_{j_1 j_2 j_3} | -\frac{\cos \theta}{m_3 R_1 R_2} | \Phi_{j'_1 j'_2 j'_3} \rangle = -\frac{1}{m_3} \langle \phi_{1 j_1} | \frac{1}{R_1} | \phi_{1 j'} \rangle \langle \phi_{2 j_2} | \frac{1}{R_2} | \phi_{2 j'_2} \rangle \langle Y_{j_3} | \cos \theta | Y_{j'_3} \rangle$$
(10)

Each of these 1-dimensional matrix elements is then evaluated using the algebraic methods described below.

A. Stretching-Coordinate Matrix Elements. As a first step, all R_i -dependent terms in the Hamiltonian of eq 1 must be expressed as functions of the Morse variable $y_i(R_i)$ of eq 8. In other words, the $v_i(R_i)^{k_i}$ potential terms in eq 3 and the R_i^{-1} and R_i^{-2} terms in the kinetic energy operator in eq 2 must be expressed in terms of the corresponding $y_i(R_i)$. This is accomplished immediately for the potential terms, ¹⁴ by choosing for each mode *i* the same value of α_i in the potential expansion terms, eq 4, and in the definition of $y_i(R_i)$, eq 8, so that $v_i \equiv y_i/(2s_i + 1) - 1$. The matrix elements of R_i^{-1} and R_i^{-2} on the basis of eq 7 could be computed exactly by numerical integration, but that approach would be contrary to the general spirit of the present method and very inefficient, since that procedure would produce a nonsparse matrix. We prefer to fit the kinetic terms with a sum of powers of v_i :

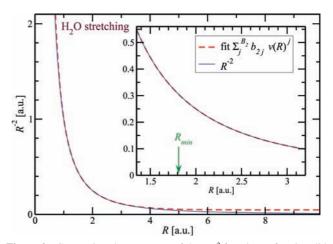


Figure 2. Comparison between one of the R_i^{-2} functions of eq 2, solid line, indicated simply as R^{-2} , and its fit in terms of Morse-related functions, eq 11, dashed line. In this calculation, relevant for the H₂O stretching coordinate *R*, the parameter $\alpha \approx 1.18 a_0^{-1}$, the parameter $R_{\min} \approx 1.81 a_0$, and the expansion extends to order $B_2 = 5$. The coefficients b_{2j} are adjusted to obtain a best fit of equally spaced points in the 1.35–3.2 a_0 range of *R* highlighted in the inset.

$$R_i^{-p} \simeq \sum_{j}^{B_{pi}} b_{pij} v_i (R_i)^j, \quad p = 1, 2$$
 (11)

An example of the quality of such a fit is illustrated in Figure 2. The fit targets a reasonably wide region around the equilibrium point $R_{i,\min}$, where it is extremely accurate, but it deteriorates especially in the large- R_i dissociation region. The fitting function could be forced to reach the same large- R_i limit as R_i^{-2} , but then the fit would deviate much more in the most important region near the minimum. Accordingly, we prefer to accept this deviation near dissociation, which causes negligible numerical error to the final spectra. The fitted R_i ranges and the best-fit coefficients b_{1ij} and b_{2ij} of R_i^{-1} and R_i^{-2} for OCS, HCN, NO₂, and H₂O are reported in Table A of the Supporting Information.²² Root mean square (rms) deviations ranging from 3 to 20 cm^{-1} are obtained for the fits based on eq 11, of the type exemplified in Figure 2. In practice these deviations affect the computed vibrational band origins at a completely negligible level.

The full details of the formalism to construct the analytical matrix elements of the functions $v_i(R_i)$ of eq 4 are reported elsewhere.^{9,23} Here we just note that the idea underlying the algebraic approach is ultimately related to supersymmetry, which provides several analytical relations for a class of exactly solvable problems, including the Schrödinger equation for the Morse potential. We only report the basic expression of the matrix elements of the exponential function and of the first derivative term:

$$\langle \phi_{ij} | e^{-\alpha_i (R_i - R_{i,\min})} | \phi_{ij} \rangle = \frac{-C_{ij'} \delta_{j,j'-1} + 2(\sigma_i + j') \delta_{j,j'} - C_{ij} \delta_{j,j'+1}}{2s_i + 1} \quad (12)$$

$$\langle \phi_{ij} | \frac{\partial}{\partial R_i} | \phi_{ij} \rangle = \frac{\alpha_i}{2} (C_{ij'} \delta_{j,j'-1} - C_{ij} \delta_{j,j'+1})$$
(13)

where $C_{ij} = [j(j + 2\sigma_i - 1)]^{1/2}$. On the basis of these expressions, all matrix elements of every stretching term in the potential expansion in eq 3 and in the kinetic terms expressed as in eq 11 can be computed exactly. The matrix representation of $v_i(R_i)^{k_i}$ is the $(2k_i + 1)$ -band diagonal.

TABLE 1: Optimized Parameters s_i and σ_i , Number of Target Bound States N_b , and Size $N_{d,i}$ of the Quantum Basis for Each Stretching Oscillator of All Molecules Studied in the Present Work

	optimized parameters				
stretching dimer	S	σ	number of bound states $N_{\rm b}$	number of basis functions $N_{d,i}$	
OC	68	40	20	30	
CS	76	52	20	30	
HC	30	0.20	20	40	
CN	74	46	20	30	
OH	26	0.01	20	30	
NO	33	5.04	20	30	

The computation of the spectrum of one pure-stretching mode through the exact diagonalization of algebraic matrices can be made substantially more efficient by choosing a properly adapted quantum mechanical basis, specifically tuned to the molecular potential.¹⁴ A better convergence of the calculation of a single oscillator improves the convergence of the full 3-dimensional spectral calculation for the triatomic molecule. For example, for the first stretching oscillator we consider

$$H_1(R_1) = -\frac{1}{2\mu_1} \frac{\partial^2}{\partial R_1^2} + \sum_{k_1=0}^{N_c} a_{k_100} v_1(R_1)^{k_1} \qquad (14)$$

The shape of the basis wave functions, eq 7, is tuned by four parameters, $R_{1,\min}$, α_1 , s_1 , and σ_1 . We fix $R_{1,\min}$ to the relevant equilibrium bond length, and to preserve the matrix sparseness, we set it to the value used in the definition of $v_1(R_1)$. By tuning the remaining s_1 and σ_1 , one has a sufficient basis flexibility to improve substantially the numerical convergence of the singleoscillator problem,¹⁴ eq 14. We optimize the parameters s_i and σ_i by the minimization of the sum of the N_b lowest bound-state eigenenergies of the dimer, as defined in eq 25 of ref 14. We optimize the energies of $N_{\rm b} = 20$ stretching bound states, which cover and far exceed the spectral range of interest. To get subcm⁻¹ accuracy, $N_{d,i} \simeq 30$ basis functions in the one-oscillator basis are usually sufficient. The values of the final optimized parameters s_i and σ_i adopted in all calculations are collected in Table 1. In fact, when $N_{d,i}$ is large enough, the spectral accuracy depends only weakly on the value of the basis parameters s_i and σ_i , so that the reported values are not particularly critical.

B. Bending-Angle Matrix Elements. On the spherical harmonics basis, eq 9, it is straightforward to express the matrix elements of powers of the $\cos \theta$ function. The bending-angle dependence of the potential energy surface is then conveniently fitted to powers of $\cos \theta - \cos \theta_{\min}$. In this basis, the bending potential matrix is sparse and can be expressed analytically. To evaluate the matrix elements of powers of $z = \cos \theta$, we use the relations²¹

$$\langle Y_{l}|z|Y_{j} \rangle = \frac{l}{\sqrt{(2j+1)(2l+1)}} \delta_{l,j+1} + \frac{j}{\sqrt{(2j+1)(2l+1)}} \delta_{l,j-1}(15)$$

$$\langle Y_{l}|z^{2}|Y_{j} \rangle = \frac{2j^{2}+2j-1}{(2j-1)(2j+3)} \delta_{l,j} + \frac{l(j+1)}{(j+l+1)\sqrt{(2j+1)(2l+1)}} \delta_{l,j+2} + \frac{j(l+1)}{(j+l+1)\sqrt{(2j+1)(2l+1)}} \delta_{l,j-2}$$

$$(16)$$

and in general, recursively

$$\langle Y_{l}|z^{k_{3}}|Y_{j'}\rangle = \frac{j+1}{\sqrt{(2j+1)(2j+3)}} \langle Y_{l}|z^{k_{3}-1}|Y_{j+1}\rangle + \frac{j}{\sqrt{(2j+1)(2j-1)}} \langle Y_{l}|z^{k_{3}-1}|Y_{j-1}\rangle$$
(17)

Like for stretching matrix elements, the matrix representing $u(\theta)^{k_3}$ is the $(2k_3 + 1)$ -band diagonal.

The derivation of the kinetic bending matrix elements is described in the Appendix. The final result is the following simple tridiagonal expression:

$$\langle Y_l | T_{\text{bend}} | Y_j \rangle = \frac{1}{2} \left(\frac{1}{\mu_1 R_1^2} + \frac{1}{\mu_2 R_2^2} \right) j(j+1) \delta_{l,j} - \frac{1}{m_3 R_1 R_2} \frac{1}{\sqrt{(2l+1)(2j+1)}} (l^3 \delta_{l,j+1} + j^3 \delta_{l,j-1}) \quad (18)$$

The basis functions in eq 9 are independent of the shape of the bending potential, which means that, for bent molecules in particular, the number of basis functions required by a given degree of convergence could become unreasonably large. Unfortunately, there is no way of optimizing the individual functions to any given angular potential, as we do for the stretching functions. One way to overcome this problem is to replace the angular functions in eq 9 with suitably optimized linear combinations thereof, for example, obtained as low-energy eigenstates of a purely bending 1-dimensional problem based on T_{bend} plus the a_{00k_3} part of the potential expansion in eq 3, as was suggested, e.g., in ref 15 and references therein. Such a kind of approach can lead to significant basis size reduction, but also, unfortunately, to entirely nonzero angular matrix elements, eventually leading to a less sparse total matrix of *H*.

III. Results

We come to illustrate the application of the proposed algebraic method to the calculation of the vibrational spectra of real triatomic molecules. We target mainly the purely vibrational energy levels (J = 0). As trial systems we select two linear molecules, OCS and HCN, and two bent ones, NO₂ and H₂O. They are all well studied by experimental as well as theoretical techniques which provide accurate spectra for comparison. In particular, for OCS, NO₂, and H₂O the literature offers realistic PESs parametrized in the form of eq 3 suitable for our method.^{18,19,24} Accordingly, we have taken the PES function parameters $a_{k_1k_2k_3}$ for OCS, NO₂, and H₂O exactly as in the referred publications.²⁵ For HCN, we use the potential surface provided by ref 26 to fit an expansion of the type in eq 3. All these PESs are determinated starting from ab initio points, but then the function parameters $a_{k_1k_2k_3}$ are adjusted so that the deviation between the calculated vibrational levels and the observed spectra is minimized. In the energy region considered, these PESs are accurate enough to allow one to predict even highly excited vibrational levels with great precision.

TABLE 2: Values of Several Relevant Potential Parameters for the Molecules Studied in the Present Work^a

system	$R_{1,\min}(a_0)$	$R_{2,\min}(a_0)$	θ_{\min} (deg)	$\alpha_1 (a_0^{-1})$	$\alpha_2 (a_0^{-1})$	ref
OCS	2.1849 (OC)	2.9506 (CS)	180.0	1.2382	1.0318	18
HCN	2.0135 (HC)	2.1793 (CN)	180.0	0.9727	1.2290	26
NO_2	2.2435	2.2435	133.767	1.6853	1.6853	24
H_2O	1.8112	1.8112	104.440	1.1769	1.1769	19

^a The HCN parameters are obtained through a best fit to 737 points on the PES as computed in ref 26.

TABLE 3: Basis Size for the Individual Oscillators $N_{d,i}$, Total Basis Size N_d , and Standard Deviations of the Computed Spectra from the Available Experimental Vibrational Levels and from Previous Calculations of the Same Levels

	number of basis functions			functions			rms dev	iation
system	$N_{\rm d,1}$	$N_{\rm d,2}$	$N_{\rm d,3}$	N _d	number of energy levels	up to energy (cm ⁻¹)	obsd - calcd (cm ⁻¹)	$obsd - ref (cm^{-1})$
OCS	30	30	50	45×10^{3}	145	8057	0.31	0.2618
HCN	40	30	50	60×10^{3}	34	12389	14	12^{26}
NO_2	30	30	55	49.5×10^{3}	143	8979	1.7	2.1^{24}
H_2O	30	30	50	45×10^3	69	21247	5.6	1.2^{19}

The values of the Morse exponential parameters α_i and equilibrium positions $R_{i,\min}$ are collected in Table 2. We fit the algebraic potential parameters $a_{k_1k_2k_3}$ for HCN to 737 points obtained using the PES function as given in ref 26, restricted to the HCN side of the HCN–HNC isomerization transition. The angular range is $\theta = 180-90^\circ$, and the coordinate ranges are $R_{\rm HC}/a_0 = 1.4-4.4$ and $R_{\rm CN}/a_0 = 1.6-3.0$, with energy up to 45 000 cm⁻¹. Table B in the Supporting Information²² reports the resulting best fit parameters $a_{k_1k_2k_3}$, which produce an rms deviation of 29 cm⁻¹.

As the bending frequency is usually smaller than the stretching frequency, and as the bending basis in eq 9 cannot be optimized to the problem at hand, it is generally necessary to include a larger number $N_{d,3}$ of bending basis states. The minimum size $N_d = N_{d,1}N_{d,2}N_{d,3}$ of the matrix for the total Hamiltonian, eq 1, which needs to be diagonalized for a very accurate and reasonably stable spectrum is eventually very moderate, on the order of $N_d \approx 10^4$. The nonzero matrix elements of *H* are on the order of $\sim [^2/_3 \max(N_c, B_{11}, B_{12}, B_{21}, B_{22}) + 1]^3$ in each row of the global matrix. For $B_{ki} = 5 \ge N_c$, as in most calculations of the present work, this yields approximately 80 nonzero matrix elements per row, i.e., a very sparse matrix to store and diagonalize.

Once the full matrix is constructed and stored, it is diagonalized using the Jacobi-Davidson method provided by the PRIMME package^{27,28} for sparse matrices. Table 3 summarizes the quality of the computed spectra, in terms of rms deviations between the observed and calculated vibrational energy levels. These levels cover an energy range extending from the ground state up to several thousand wavenumbers, as indicated in Table 3. The good agreement with experiment shown by the small rms deviations indicates both the high quality of the considered PES parametrizations and the satisfactory accuracy of the employed method. The complete spectral levels obtained by means of the present approach and their detailed comparison to experimental data for OCS, HCN, NO2, and H2O are collected in Tables C-F in the Supporting Information.²² For OCS we compute and compare to experiment also numerous Π (m = ± 1) states.

For the molecules for which spectra were computed before on the basis of the same PES (but with a different approach to the solution of the quantum mechanical problem), we obtain an essentially equivalent accuracy of the spectra, as shown by the similar rms deviations reported in Table 3. Basically all discrepancies with experiment are to be attributed to the lack of accuracy of the adopted PES. Table D in the Supporting Information²² also collects a few predicted energy levels of HCN in the 7000–12000 cm⁻¹ spectral range. These calculated values are also in good agreement with previous calculations by Mourik et al. Table F in the Supporting Information²² also reports a few predicted levels for H_2O .

The level assignments in terms of local vibrations is not always straightforward. The wave function of each vibrational state can be analyzed, and it always results in a complicated admixture of excitations of all three local vibrational modes. For instance, the OCS vibrational excitation at $E_v = 2937.2$ cm⁻¹ consists mainly (35%) of v = (140) plus 21% v = (060), plus other minor components, which confirms the traditional assignment (140) reported in Table C in the Supporting Information. However, for example, the largest local component, v = (080), of the OCS vibrational excitation near 3990 cm⁻¹ is 26%, while the component on its standard local assignment, v = (160), is smaller (18%).

IV. Discussion and Conclusions

In the present work, we demonstrate the possibility to compute the vibrational spectrum of an arbitrary triatomic molecule on the basis of its PES and using algebraic techniques for the analytical determination of the matrix elements. The advantages of this method include simple formulas for the matrix elements, moderate total Hilbert space size, and sparseness of the Hamiltonian matrix, all cooperating to a fast and efficient diagonalization. The generalization of this method to four-atomic and larger molecules is in principle straightforward, and in these higher-dimensional contexts the reduced basis size for each degree of freedom is even more crucial. These advantages make this method a very promising tool for the analysis of the vibrational levels of small molecules.

The input of the present approach includes only the atomic masses and a reasonably dense numerical sampling of the PES: this is in principle within reach of ab initio electronic structure quantum chemical calculations. We did carry out a calculation of the spectrum of HCN based on the ab initio PES generated by a large set of DFT-LDA (density functional theory in the local-density approximation) calculations. Due to the known drawbacks of the LDA, the resulting spectrum shows deviations from the experimental levels of a few hundred cm⁻¹, but it proves the possibility to compute even highly excited vibrational molecular states entirely from first principles, with no parameters adjusted to the observed spectroscopic data. When highly accurate quantum chemical methods are applied to a relatively

fine and extensive determination of the PES of a triatomic molecule, it will be possible to obtain much better predictive power for a calculation entirely free from any experimental input.

If the PES is known to a high degree of accuracy even in the energy region near and slightly above dissociation, the present method can take advantage of the possibility to extend the basis set beyond the number of bound states to describe reliably weakly bound predissociation states in the quasi-continuum. In the future, the present algebraic approach could be extended to the study of resonances in the continuum, bound-to-free transitions in infrared absorption, Franck-Condon processes, and in principle even atom-molecule and molecule-molecule collisions. Other perspective applications include the area of nonrigid molecules (van der Waals complexes, quasi-linear molecules)¹¹ and potentials with many minima such as those occurring in torsional oscillations.¹² Note finally that a reliable description of the quantum vibrational dynamics could allow us to exploit the manageable Hamiltonian structure produced by the present method to the study of intramolecular vibrational energy redistribution.

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Appendix: Angular Matrix Elements

The kinetic energy operator, eq 2, contains terms proportional to $\cot \theta$ and to $[\cos(\theta/2)]^{-2}$, which are singular at the boundary points $\theta = 0$ and π . In fact, $\theta = \pi$ is an especially important configuration for linear molecules, and it should then be described smoothly. Luckily, since the integration implied in the matrix elements is performed over the variable $z = \cos \theta$, this divergence is not a problem, and all relevant matrix elements can be computed analytically using several properties of the spherical harmonics. For brevity, we replace the θ -independent kinetic coefficients with the shorthand

$$a_1 = -\frac{1}{2} \left(\frac{1}{\mu_1 R_1^2} + \frac{1}{\mu_2 R_2^2} \right)$$
(A1)

$$a_2 = \frac{1}{m_3 R_1 R_2}$$
 (A2)

and define the pure bending part of T as

$$T_{\text{bend}}(\theta) = (a_1 + a_2 \cos \theta) \left(\frac{\partial^2}{\partial \theta^2} + \cot \theta \frac{\partial}{\partial \theta} \right) - a_2 \left(\sin \theta \frac{\partial}{\partial \theta} + \cos \theta \right) + \frac{a_2 - a_1}{2} \frac{1}{1 + \cos \theta} \Pi_z^2 \quad (A3)$$

In terms of $z = \cos \theta$, $\partial/\partial \theta \rightarrow -[\sin \theta(\partial/(\partial \cos \theta))] = -(1 - z^2)^{1/2}(\partial/\partial z)$, and $\partial^2/\partial \theta^2 \rightarrow \sin^2 \theta(\partial^2/(\partial \cos^2 \theta)) - \cos \theta(\partial/(\partial \cos \theta)) = (1 - z^2)(\partial^2/\partial z^2) - z(\partial/\partial z)$, the angular kinetic energy is conveniently decomposed as

$$T_{\text{bend}}(z) = a_1(1-z^2)\frac{\partial^2}{\partial z^2} \rightarrow T_1$$

+ $a_2 z(1-z^2)\frac{\partial^2}{\partial z^2} \rightarrow T_2$
+ $a_1 \left(-2z \frac{\partial}{\partial z}\right) \rightarrow T_3$
+ $a_2 \left(-2z^2 \frac{\partial}{\partial z}\right) \rightarrow T_4$
+ $a_2(1-z^2)\frac{\partial}{\partial z} \rightarrow T_5$
- $a_2 z \rightarrow T_6$
+ $\frac{a_2 - a_1}{2} \frac{1}{1+z}m^2 \rightarrow T_7$ (A4)

We derive the matrix elements of T_{bend} in the general case of arbitrary integer *m*. Because for the $m \neq 0$ "II", " Δ ", etc. vibrational excitations of linear molecules the T_7 term is singular at $\theta = 180^\circ$, all its matrix elements in the Y_{l0} basis diverge: for general *m* we need then to consider the natural extension of the basis of eq 9, namely, the one provided by the Y_{lm} spherical harmonics:

$$Y_{l}^{m}(\theta) = \sqrt{2\pi}Y_{lm}(\theta, 0) = N_{lm}P_{l}^{m}(\cos\theta)$$

with $N_{lm} = \left[\frac{(2l+1)(l-|m|)!}{2(l+|m|)!}\right]^{1/2}$ (A5)

for $l \ge |m|$. For compactness, in the following we write the matrix elements in terms of associated Legendre polynomials $P_l^m(z) \equiv (-1)^{(lml+m)/2}(1-z^2)^{lml/2} (dP_l(z)/dz^{lml})$; i.e., we omit the normalization factor N_{lm} , and we use the notation $P'_l^m = (d/dz)P_l^m$. In the derivation of the T_{bend} matrix elements, we make use of the following identities for the associated Legendre polynomials:

$$(j + 1 - |m|)P_{j+1}^{m}(z) = (2j + 1)zP_{j}(z) - (j + |m|)P_{j-1}^{m}(z)$$
(A6)
$$(1 - z^{2})P_{j}^{m}(z) = -izP_{j}^{m}(z) + (i + |m|)P_{j}^{m}(z) - (A7)$$

$$(1-z)P'_{j}(z) = -jzP'_{j}(z) + (j+m)P_{j-1}(z) \quad (A7)$$

$$(1-z^{2})P''_{j}(z) = 2zP''_{j}(z) - \left[j(j+1) - \frac{m^{2}}{1-z^{2}}\right]P_{j}^{m}(z) \quad (A8)$$

As a direct consequence of eq A6, eq 15 generalizes to

$$\langle Y_{l}^{m}|z|Y_{j}^{m}\rangle = \frac{N_{jm}}{N_{lm}} \left(\frac{l-|m|}{2j+1}\delta_{l,j+1} + \frac{j+|m|}{2j+1}\delta_{l,j-1}\right)$$
(A9)

and the recursive relation 17 generalizes to

. .

$$\langle Y_l^m | z^{k_3} | Y_j^m \rangle = \frac{N_{jm}}{2j+1} \left(\frac{j+1-|m|}{N_{j+1m}} \langle Y_l^m | z^{k_3-1} | Y_{j+1}^m \rangle \right) + \frac{j+|m|}{N_{j-1m}} \langle Y_l | z^{k_3-1} | Y_{j-1} \rangle$$
 (A10)

These relations are also useful for the calculation of the potential matrix elements.

The individual kinetic terms are then expressed as

$$\langle P_{l}^{m} | T_{2} | P_{j}^{m} \rangle = a_{2} \langle P_{l}^{m} | z(1 - z^{2}) | P_{j}^{\prime \prime m} \rangle$$

$$= a_{2} \langle P_{l}^{m} | 2z^{2} | P_{j}^{\prime m} \rangle - a_{2} j(j + 1) \langle P_{l}^{m} | z | P_{j}^{m} \rangle +$$

$$a_{2} m^{2} \langle P_{l}^{m} | \frac{z}{1 - z^{2}} | P_{j}^{m} \rangle$$
(A12)

$$\langle P_l^m | T_3 | P_j^m \rangle = -a_1 \langle P_l^m | 2z | P_j'^m \rangle \tag{A13}$$

$$\langle P_l^m | T_4 | P_j^m \rangle = -a_2 \langle P_l^m | 2z^2 | P_j'^m \rangle \tag{A14}$$

$$\langle P_l^m | T_5 | P_j^m \rangle = a_2 \langle P_l^m | (1 - z^2) | P_j^m \rangle = -a_2 j \langle P_l^m | z | P_j^m \rangle + a_2 (j + |m|) \langle P_l^m | P_{j-l}^m \rangle$$
 (A15)

$$\langle P_l^m | T_6 | P_j^m \rangle = -a_2 \langle P_l^m | z | P_j^m \rangle \tag{A16}$$

$$\langle P_l^m | T_7 | P_j^m \rangle = \frac{a_2 - a_1}{2} m^2 \langle P_l^m | \frac{1}{1 + z} | P_j^m \rangle$$
 (A17)

By collecting everything together and simplifying, we obtain

$$\langle P_l^m | T_{\text{bend}} | P_j^m \rangle = -a_1 j (j+1) \langle P_l^m | P_j^m \rangle - a_2 (j+1)^2 \langle P_l^m | z | P_j^m \rangle + a_2 (j+|m|) \langle P_l^m | P_{j-1}^m \rangle + \frac{a_1 + a_2}{2} m^2 \langle P_l^m | \frac{1}{1-z} | P_j^m \rangle$$
 (A18)

Accordingly, the bending kinetic matrix elements in the normalized basis Y_l^m are

$$\begin{split} \langle Y_{l}^{m}|T_{\text{bend}}|Y_{j}^{m}\rangle &= -a_{1}j(j+1)\delta_{lj} - a_{2}(j+1)^{2}\langle Y_{l}|z|Y_{j}\rangle + \\ a_{2}(j+|m|)\frac{N_{jm}}{N_{j-1m}}\delta_{l,j-1} + \frac{a_{1}+a_{2}}{2}m^{2}\langle Y_{l}^{m}|\frac{1}{1-z}|Y_{j}^{m}\rangle \\ &= -a_{1}j(j+1)\delta_{l,j} - \\ \frac{a_{2}}{\sqrt{(2j+1)(2l+1)}}[l^{2}\sqrt{l^{2}-m^{2}}\delta_{l,j+1} + j^{2}\sqrt{j^{2}-m^{2}}\delta_{l,j-1}] + \\ &= \frac{a_{1}+a_{2}}{2}m^{2}\langle Y_{l}^{m}|\frac{1}{1-z}|Y_{j}^{m}\rangle \quad (A19) \end{split}$$

where we have inserted the matrix elements of z from eq A9 and combined the two non-Hermitian terms proportional to a_2 into a final explicitly Hermitian expression. The important special case m = 0 is reported as eq 18 in the main text. The matrix element in the final term is

$$\langle Y_l^m | \frac{1}{1-z} | Y_j^m \rangle = N_{lm} N_{jm} l_{\max} [l_{\max} + 1] K_m (l_{\max})$$

with $l_{\max} = \max(l, j)$ (A20)

and with

$$K_{\pm 1}(l) = 1$$

$$K_{\pm 2}(l) = \frac{1}{2}(l^2 + l - 2)$$

$$K_{\pm 3}(l) = \frac{1}{3}(l^4 + 2l^3 - 7l^2 - 8l + 12)$$

Unfortunately the last term in eq A19, the one proportional to m^2 and originated by T_7 plus parts of T_1 and T_2 , produces a

matrix which, according to eq A20, is not sparse on this basis. This leads to significant numerical overhead in the calculation of the $m \neq 0$ states, with respect to the m = 0 states.

Supporting Information Available: Best-fit parameters describing the PES of HCN according to eq 3, b_{pik} best-fit parameters of eq 11 for the R_i^{-p} functions, detailed tables of the computed vibrational band origins compared to experimental data, and for HCN and H₂O a comparison with spectral data calculated using different approaches. This material is available free of charge via the Internet at http://pubs.acs.org.

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